H

National Qualifications 2015

X757/76/02

**Physics
Section 1—Questions**

TUESDAY, 5 MAY
1:00 PM – 3:30 PM

Instructions for the completion of Section 1 are given on *Page two* of your question and answer booklet X757/76/01.

Record your answers on the answer grid on *Page three* of your question and answer booklet.

Reference may be made to the Data Sheet on *Page two* of this booklet and to the Relationships Sheet X757/76/11.

Before leaving the examination room you must give your question and answer booklet to the Invigilator; if you do not, you may lose all the marks for this paper.

X7577602

DATA SHEET

COMMON PHYSICAL QUANTITIES

Quantity	Symbol	Value	Quantity	Symbol	Value
Speed of light in vacuum	c	$3{\cdot}00 \times 10^8\,\text{m s}^{-1}$	Planck's constant	h	$6{\cdot}63 \times 10^{-34}\,\text{J s}$
Magnitude of the charge on an electron	e	$1{\cdot}60 \times 10^{-19}\,\text{C}$	Mass of electron	m_e	$9{\cdot}11 \times 10^{-31}\,\text{kg}$
Universal Constant of Gravitation	G	$6{\cdot}67 \times 10^{-11}\,\text{m}^3\,\text{kg}^{-1}\,\text{s}^{-2}$	Mass of neutron	m_n	$1{\cdot}675 \times 10^{-27}\,\text{kg}$
Gravitational acceleration on Earth	g	$9{\cdot}8\,\text{m s}^{-2}$	Mass of proton	m_p	$1{\cdot}673 \times 10^{-27}\,\text{kg}$
Hubble's constant	H_0	$2{\cdot}3 \times 10^{-18}\,\text{s}^{-1}$			

REFRACTIVE INDICES

The refractive indices refer to sodium light of wavelength 589 nm and to substances at a temperature of 273 K.

Substance	Refractive index	Substance	Refractive index
Diamond	2·42	Water	1·33
Crown glass	1·50	Air	1·00

SPECTRAL LINES

Element	Wavelength/nm	Colour	Element	Wavelength/nm	Colour
Hydrogen	656	Red	Cadmium	644	Red
	486	Blue-green		509	Green
	434	Blue-violet		480	Blue
	410	Violet			
	397	Ultraviolet		Lasers	
	389	Ultraviolet	Element	Wavelength/nm	Colour
			Carbon dioxide	9550 } 10590 }	Infrared
Sodium	589	Yellow	Helium-neon	633	Red

PROPERTIES OF SELECTED MATERIALS

Substance	Density/kg m^{-3}	Melting Point/K	Boiling Point/K
Aluminium	$2{\cdot}70 \times 10^3$	933	2623
Copper	$8{\cdot}96 \times 10^3$	1357	2853
Ice	$9{\cdot}20 \times 10^2$	273	. . . .
Sea Water	$1{\cdot}02 \times 10^3$	264	377
Water	$1{\cdot}00 \times 10^3$	273	373
Air	1·29	. . . .	. . . .
Hydrogen	$9{\cdot}0 \times 10^{-2}$	14	20

The gas densities refer to a temperature of 273 K and a pressure of $1{\cdot}01 \times 10^5$ Pa.

1. The following velocity-time graph represents the vertical motion of a ball.

velocity (m s^{-1})
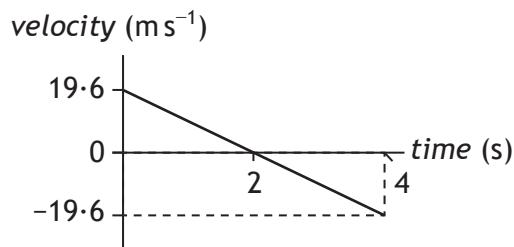
Which of the following acceleration-time graphs represents the same motion?

A   *acceleration* (m s^{-2})
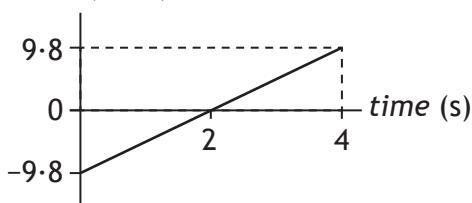
B   *acceleration* (m s^{-2})
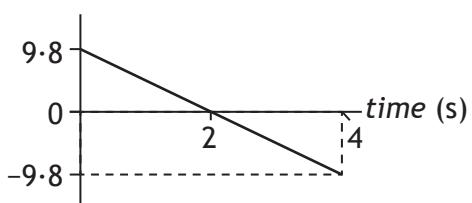
C   *acceleration* (m s^{-2})
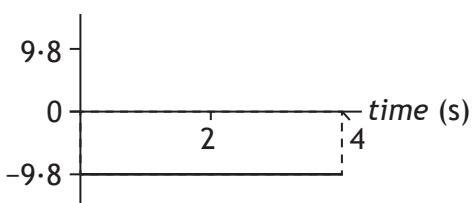
D   *acceleration* (m s^{-2})
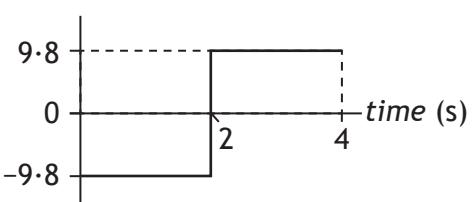
E   *acceleration* (m s^{-2})
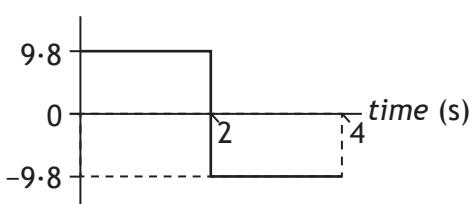
2. A car is travelling at $12\,\mathrm{m\,s^{-1}}$ along a straight road. The car now accelerates uniformly at $-1\cdot5\,\mathrm{m\,s^{-2}}$ for $6\cdot0\,\mathrm{s}$.

The distance travelled during this time is

A $18\,\mathrm{m}$

B $45\,\mathrm{m}$

C $68\,\mathrm{m}$

D $72\,\mathrm{m}$

E $99\,\mathrm{m}$.

3. A box of mass m rests on a slope as shown.
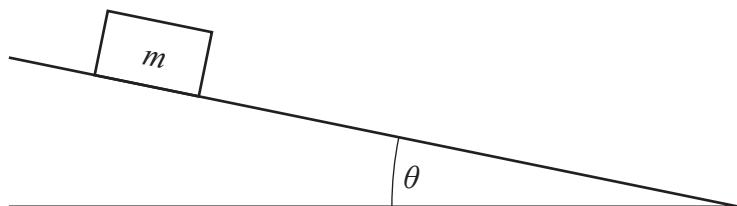
Which row in the table shows the component of the weight acting down the slope and the component of the weight acting normal to the slope?

|	Component of weight acting down the slope	Component of weight acting normal to the slope
A	$mg\,\sin\theta$	$mg\,\cos\theta$
B	$mg\,\tan\theta$	$mg\,\sin\theta$
C	$mg\,\cos\theta$	$mg\,\sin\theta$
D	$mg\,\cos\theta$	$mg\,\tan\theta$
E	$mg\,\sin\theta$	$mg\,\tan\theta$

4. A person stands on bathroom scales in a lift.

The scales show a reading greater than the person's weight.

The lift is moving

A   upwards with constant speed

B   downwards with constant speed

C   downwards with increasing speed

D   downwards with decreasing speed

E   upwards with decreasing speed.

5. A car of mass 900 kg pulls a caravan of mass 400 kg along a straight, horizontal road with an acceleration of $2 \cdot 0\,\text{m s}^{-2}$.


coupling

900 kg

400 kg
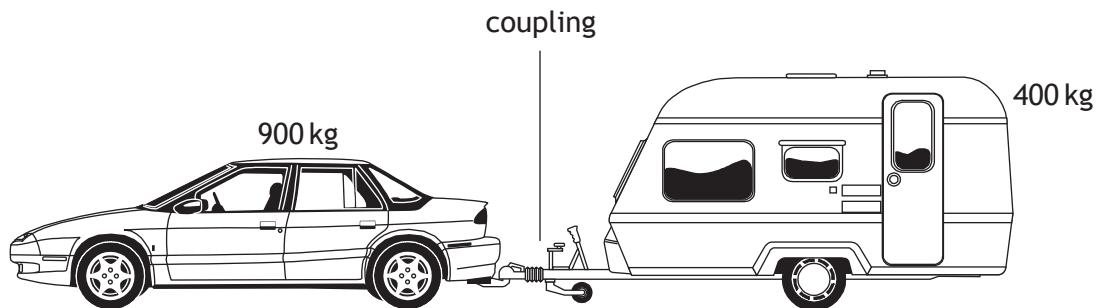
Assuming that the frictional forces on the caravan are negligible, the tension in the coupling between the car and the caravan is

A 400 N

B 500 N

C 800 N

D 1800 N

E 2600 N.

6. Water flows at a rate of $6 \cdot 25 \times 10^8\,\text{kg}$ per minute over a waterfall.

The height of the waterfall is 108 m.

The total power delivered by the water in falling through the 108 m is

A $1 \cdot 13 \times 10^9\,\text{W}$

B $1 \cdot 10 \times 10^{10}\,\text{W}$

C $6 \cdot 62 \times 10^{11}\,\text{W}$

D $4 \cdot 05 \times 10^{12}\,\text{W}$

E $3 \cdot 97 \times 10^{13}\,\text{W}$.

7. A spacecraft is travelling at a constant speed of $0 \cdot 60c$ relative to the Moon.

An observer on the Moon measures the length of the moving spacecraft to be 190 m.

The length of the spacecraft as measured by an astronaut on the spacecraft is

A 120 m

B 152 m

C 238 m

D 297 m

E 300 m.

[Turn over

8. A siren on an ambulance emits sound at a constant frequency of 750 Hz.

   The ambulance is travelling at a constant speed of 25·0 m s⁻¹ towards a stationary observer.

   The speed of sound in air is 340 m s⁻¹.

   The frequency of the sound heard by the observer is

   A   695 Hz

   B   699 Hz

   C   750 Hz

   D   805 Hz

   E   810 Hz.

9. The emission of beta particles in radioactive decay is evidence for the existence of

   A   quarks

   B   electrons

   C   gluons

   D   neutrinos

   E   bosons.

10. Two parallel metal plates X and Y in a vacuum have a potential difference V across them.
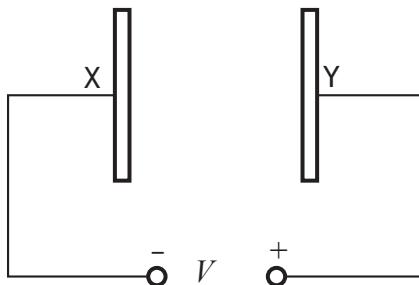
   An electron of charge e and mass m, initially at rest, is released from plate X.

   The speed of the electron when it reaches plate Y is given by

   A $\dfrac{2eV}{m}$

   B $\sqrt{\dfrac{2eV}{m}}$

   C $\sqrt{\dfrac{2V}{em}}$

   D $\dfrac{2V}{em}$

   E $\dfrac{2mV}{e}$

11. A potential difference of 2 kV is applied across two metal plates.

An electron passes between the metal plates and follows the path shown.
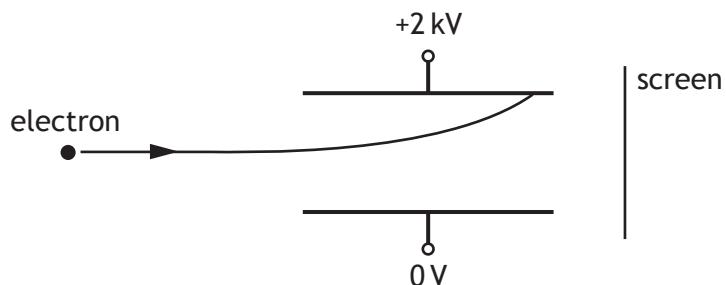
A student makes the following statements about changes that could be made to allow the electron to pass between the plates and reach the screen.

  I Increasing the initial speed of the electron could allow the electron to reach the screen.

  II Increasing the potential difference across the plates could allow the electron to reach the screen.

  III Reversing the polarity of the plates could allow the electron to reach the screen.

Which of these statements is/are correct?

A   I only

B   II only

C   III only

D   I and II only

E   I and III only

12. The following statement describes a fusion reaction.

$$^{2}_{1}H + {}^{2}_{1}H \rightarrow {}^{3}_{2}He + {}^{1}_{0}n + energy$$
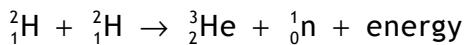
The total mass of the particles before the reaction is $6{\cdot}684 \times 10^{-27}$ kg.

The total mass of the particles after the reaction is $6{\cdot}680 \times 10^{-27}$ kg.

The energy released in the reaction is

A $6{\cdot}012 \times 10^{-10}$ J

B $6{\cdot}016 \times 10^{-10}$ J

C $1{\cdot}800 \times 10^{-13}$ J

D $3{\cdot}600 \times 10^{-13}$ J

E $1{\cdot}200 \times 10^{-21}$ J.

13. Two identical loudspeakers, L_1 and L_2, are operated at the same frequency and in phase with each other. An interference pattern is produced.
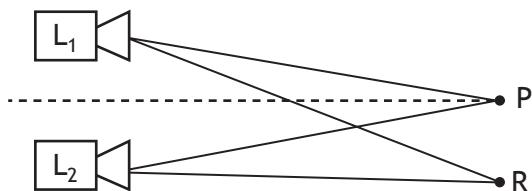
At position P, which is the same distance from both loudspeakers, there is a maximum.

The next maximum is at position R, where $L_1R = 5\cdot6$ m and $L_2R = 5\cdot3$ m.

The speed of sound in air is $340\,\text{m}\,\text{s}^{-1}$.

The frequency of the sound emitted by the loudspeakers is

A $8\cdot8 \times 10^{-4}\,\text{Hz}$

B $3\cdot1 \times 10^{1}\,\text{Hz}$

C $1\cdot0 \times 10^{2}\,\text{Hz}$

D $1\cdot1 \times 10^{3}\,\text{Hz}$

E $3\cdot7 \times 10^{3}\,\text{Hz}.$

14. An experiment is carried out to measure the wavelength of red light from a laser.

The following values for the wavelength are obtained.

650 nm   640 nm   635 nm   648 nm   655 nm

The mean value for the wavelength and the approximate random uncertainty in the mean is

A   (645 ± 1) nm

B   (645 ± 4) nm

C   (646 ± 1) nm

D   (646 ± 4) nm

E   (3228 ± 20) nm.

**15.** Red light is used to investigate the critical angle of two materials P and Q.
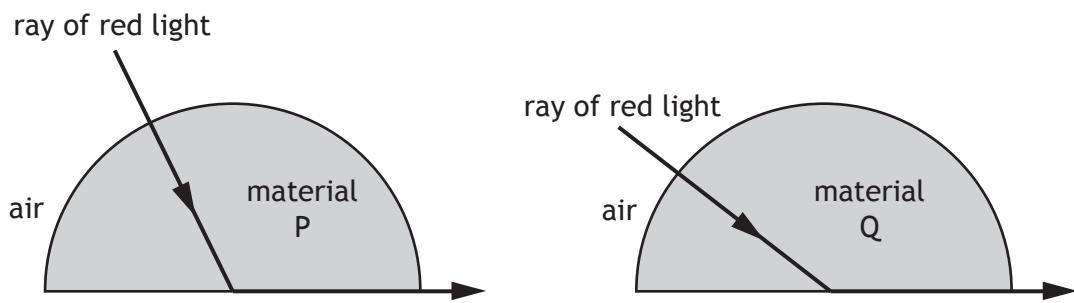
A student makes the following statements.

  I Material P has a higher refractive index than material Q.

  II The wavelength of the red light is longer inside material P than inside material Q.

  III The red light travels at the same speed inside materials P and Q.

Which of these statements is/are correct?

A   I only

B   II only

C   III only

D   I and II only

E   I, II and III

16. The diagram represents some electron transitions between energy levels in an atom.
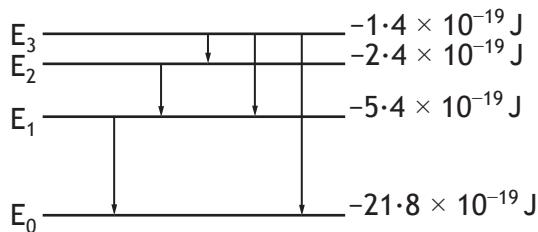

The radiation emitted with the shortest wavelength is produced by an electron making transition

A E_1 to E_0

B E_2 to E_1

C E_3 to E_2

D E_3 to E_1

E E_3 to E_0.

[Turn over

17. The output from a signal generator is connected to the input terminals of an oscilloscope. The trace observed on the oscilloscope screen, the Y-gain setting and the timebase setting are shown.
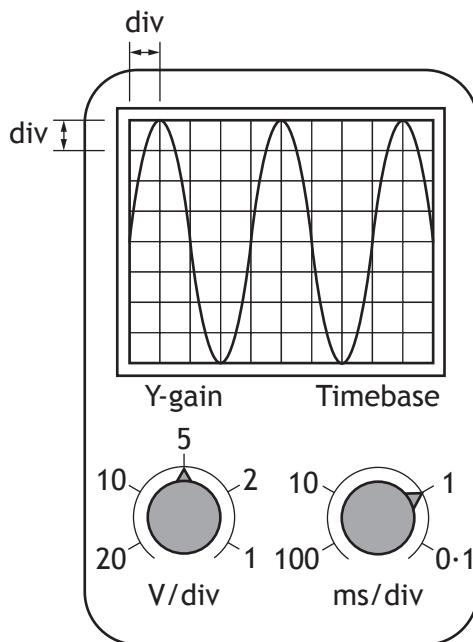
The frequency of the signal shown is calculated using the

A   timebase setting and the vertical height of the trace

B   timebase setting and the horizontal distance between the peaks of the trace

C   Y-gain setting and the vertical height of the trace

D   Y-gain setting and the horizontal distance between the peaks of the trace

E   Y-gain setting and the timebase setting.

18. A circuit is set up as shown.
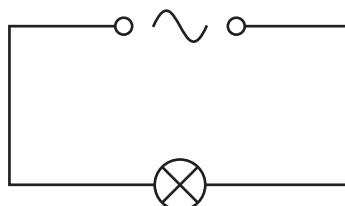
The r.m.s voltage across the lamp is 12 V.

The power produced by the lamp is 24 W.

The peak current in the lamp is

A   0·71 A

B   1·4 A

C   2·0 A

D   2·8 A

E   17 A.

19. A student makes the following statements about energy bands in different materials.

   I In metals the highest occupied energy band is not completely full.

  II In insulators the highest occupied energy band is full.

 III The gap between the valence band and conduction band is smaller in semiconductors than in insulators.

Which of these statements is/are correct?

A   I only

B   II only

C   I and II only

D   I and III only

E   I, II and III

20. The upward lift force L on the wings of an aircraft is calculated using the relationship

$$L = \tfrac{1}{2}\rho v^2 A C_L$$

where:

        ρ is the density of air
        v is the speed of the wings through the air
        A is the area of the wings
        C_L is the coefficient of lift.

The weight of a model aircraft is 80·0 N.
The area of the wings on the model aircraft is 3·0 m².
The coefficient of lift for these wings is 1·6.
The density of air is 1·29 kg m^{-3}

The speed required for the model aircraft to maintain a level flight is

A   2·5 m s^{-1}

B   3·6 m s^{-1}

C   5·1 m s^{-1}

D   12·9 m s^{-1}

E   25·8 m s^{-1}.

**[END OF SECTION 1. NOW ATTEMPT THE QUESTIONS IN SECTION 2
OF YOUR QUESTION AND ANSWER BOOKLET]**

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National Qualifications 2015

Mark

X757/76/01

Physics
Section 1 – Answer Grid and Section 2

TUESDAY, 5 MAY
1:00 PM – 3:30 PM

Fill in these boxes and read what is printed below.

Full name of centre

Town

Forename(s)

Surname

Number of seat

Date of birth

Day

Month

Year

Scottish candidate number

Total marks — 130

SECTION 1 — 20 marks
Attempt ALL questions.
Instructions for the completion of Section 1 are given on *Page two*.

SECTION 2 — 110 marks
Attempt ALL questions.

Reference may be made to the Data Sheet on *Page two* of the question paper X757/76/02 and to the Relationship Sheet X757/76/11.

Care should be taken to give an appropriate number of significant figures in the final answers to calculations.

Write your answers clearly in the spaces provided in this booklet. Additional space for answers and rough work is provided at the end of this booklet. If you use this space you must clearly identify the question number you are attempting. Any rough work must be written in this booklet. You should score through your rough work when you have written your final copy.

Use **blue** or **black** ink.

Before leaving the examination room you must give this booklet to the Invigilator; if you do not, you may lose all the marks for this paper.

SQA

PB

The questions for Section 1 are contained in the question paper X757/76/02.
Read these and record your answers on the answer grid on *Page three* opposite.
Use **blue** or **black** ink. Do NOT use gel pens or pencil.

1. The answer to each question is **either** A, B, C, D or E. Decide what your answer is, then fill in the appropriate bubble (see sample question below).

2. There is **only one correct** answer to each question.

3. Any rough work must be written in the additional space for answers and rough work at the end of this booklet.

Sample Question

The energy unit measured by the electricity meter in your home is the:

    A   ampere

    B   kilowatt-hour

    C   watt

    D   coulomb

    E   volt.

The correct answer is **B**—kilowatt-hour. The answer **B** bubble has been clearly filled in (see below).

A	B	C	D	E
○	●	○	○	○

Changing an answer

If you decide to change your answer, cancel your first answer by putting a cross through it (see below) and fill in the answer you want. The answer below has been changed to **D**.

A	B	C	D	E
○	⊗	○	●	○

If you then decide to change back to an answer you have already scored out, put a tick (✓) to the **right** of the answer you want, as shown below:

A	B	C	D	E		A	B	C	D	E
○	⊗✓	○	⊗	○	or	○	⊗✓	○	○	○

	A	B	C	D	E
1	○	○	○	○	○
2	○	○	○	○	○
3	○	○	○	○	○
4	○	○	○	○	○
5	○	○	○	○	○
6	○	○	○	○	○
7	○	○	○	○	○
8	○	○	○	○	○
9	○	○	○	○	○
10	○	○	○	○	○
11	○	○	○	○	○
12	○	○	○	○	○
13	○	○	○	○	○
14	○	○	○	○	○
15	○	○	○	○	○
16	○	○	○	○	○
17	○	○	○	○	○
18	○	○	○	○	○
19	○	○	○	○	○
20	○	○	○	○	○

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[Turn over for SECTION 2 on *Page six*

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SECTION 2 — 110 marks

Attempt ALL questions

1. The shot put is an athletics event in which competitors "throw" a shot as far as possible. The shot is a metal ball of mass 4·0 kg. One of the competitors releases the shot at a height of 1·8 m above the ground and at an angle θ to the horizontal. The shot travels through the air and hits the ground at X. The effects of air resistance are negligible.

not to scale
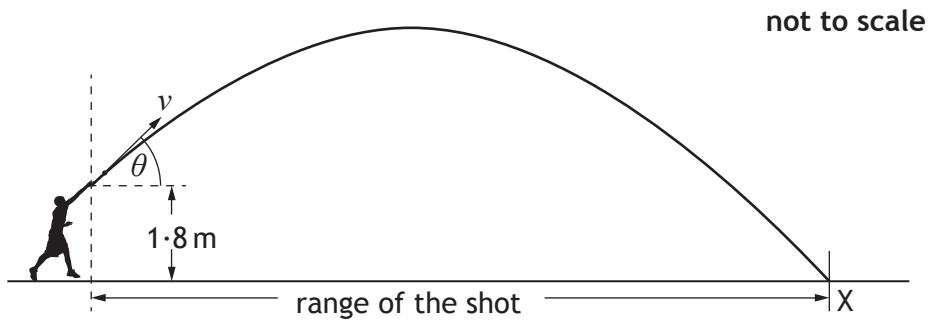
The graph shows how the release speed of the shot v varies with the angle of projection θ.
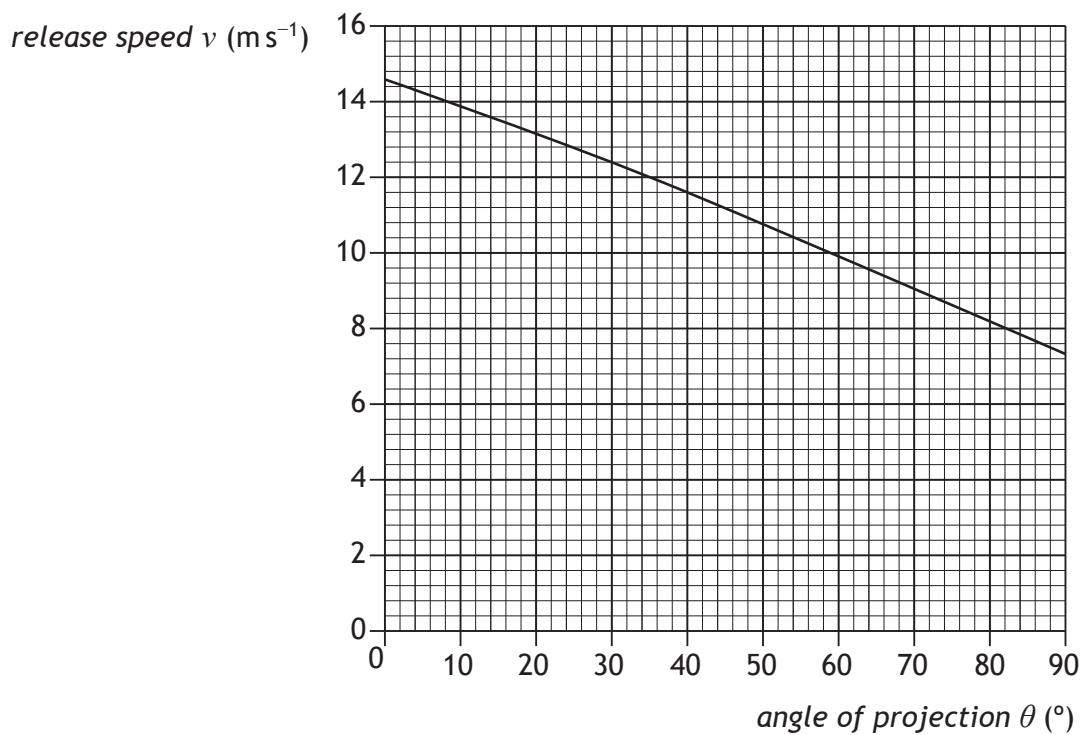
1. (continued)

(a) The angle of projection for a particular throw is 40°.

(i) (A) State the release speed of the shot at this angle.   **1**

(B) Calculate the horizontal component of the initial velocity of the shot.   **1**

Space for working and answer

(C) Calculate the vertical component of the initial velocity of the shot.   **1**

Space for working and answer

(ii) The maximum height reached by the shot is 4·7 m above the ground. The time between release and reaching this height is 0·76 s.

(A) Calculate the total time between the shot being released and hitting the ground at X.   **4**

Space for working and answer

1. (a) (ii) (continued)

        (B) Calculate the range of the shot for this throw. **3**

        Space for working and answer

  (b) Using information from the graph, explain the effect of increasing the angle of projection on the kinetic energy of the shot at release. **2**

2. A student sets up an experiment to investigate collisions between two trolleys on a long, horizontal track.
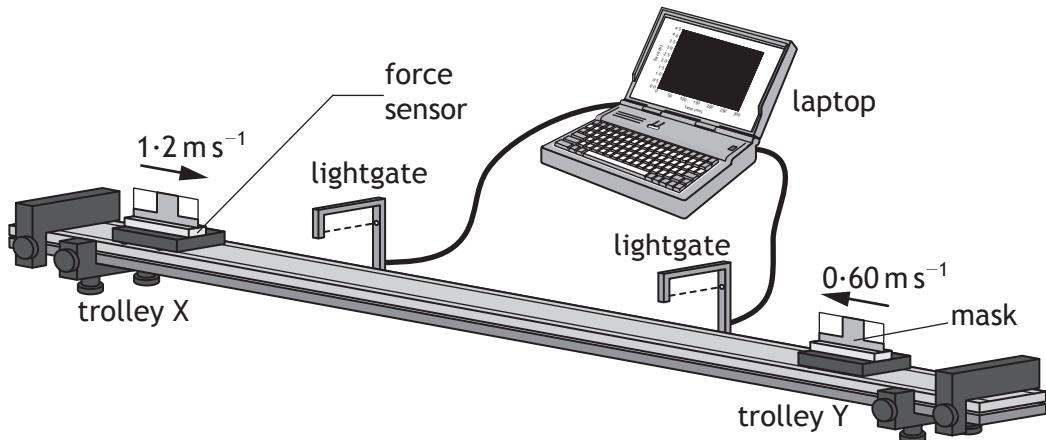
The mass of trolley X is 0·25 kg and the mass of trolley Y is 0·45 kg.

The effects of friction are negligible.

In one experiment, trolley X is moving at 1·2 m s^{-1} to the right and trolley Y is moving at 0·60 m s^{-1} to the left.

The trolleys collide and do not stick together. After the collision, trolley X rebounds with a velocity of 0·80 m s^{-1} to the left.

(a) Determine the velocity of trolley Y after the collision.

Space for working and answer

3

[Turn over

2. **(continued)**

(b) The force sensor measures the force acting on trolley Y during the collision.

The laptop displays the following force-time graph for the collision.
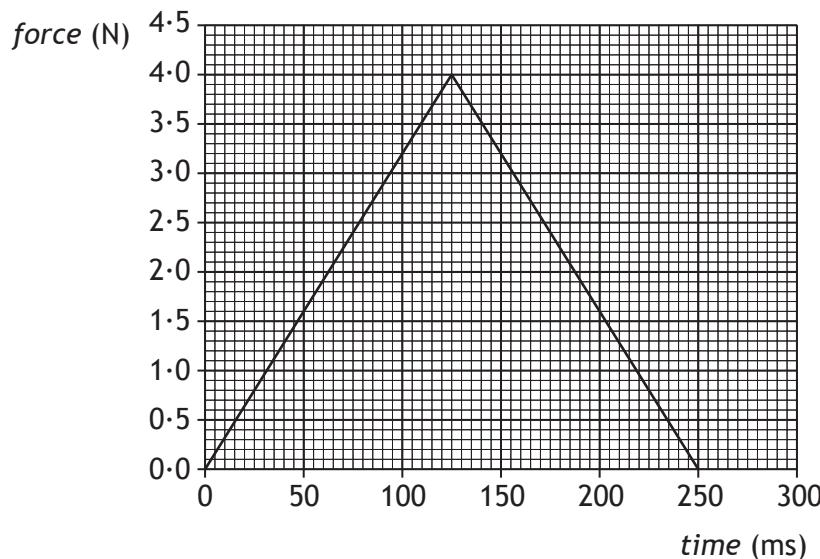
(i) Determine the magnitude of the impulse on trolley Y. **3**

Space for working and answer

(ii) Determine the magnitude of the change in momentum of trolley X. **1**

2. (b) (continued)

(iii) Sketch a velocity-time graph to show how the velocity of trolley X varies from 0·50 s before the collision to 0·50 s after the collision. **3**

Numerical values are required on both axes.
You may wish to use the square-ruled paper on *Page thirty-six*.

[Turn over

3. A space probe of mass $5{\cdot}60 \times 10^3$ kg is in orbit at a height of $3{\cdot}70 \times 10^6$ m above the surface of Mars.
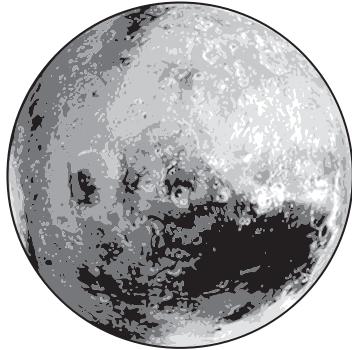
space probe

not to scale

Mars

The mass of Mars is $6{\cdot}42 \times 10^{23}$ kg.
The radius of Mars is $3{\cdot}39 \times 10^6$ m.

(a) Calculate the gravitational force between the probe and Mars. **3**

Space for working and answer

(b) Calculate the gravitational field strength of Mars at this height. **3**

Space for working and answer

[Turn over for Question 4 on *Page fourteen*

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4. Light from the Sun is used to produce a visible spectrum.

A student views this spectrum and observes a number of dark lines as shown.

wavelength (nm)
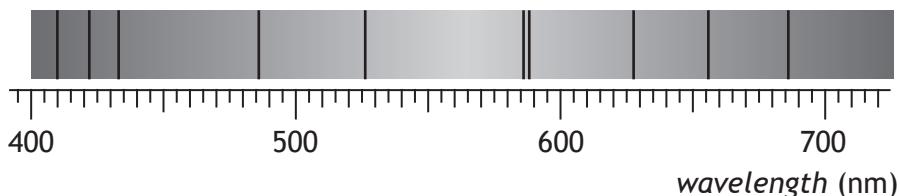
(a) Explain how these dark lines in the spectrum of sunlight are produced.　　**2**

(b) One of the lines is due to hydrogen.

The position of this hydrogen line in the visible spectrum is shown for a distant galaxy, a nearby galaxy and the Sun.

distant galaxy

nearby galaxy

the Sun

wavelength (nm)
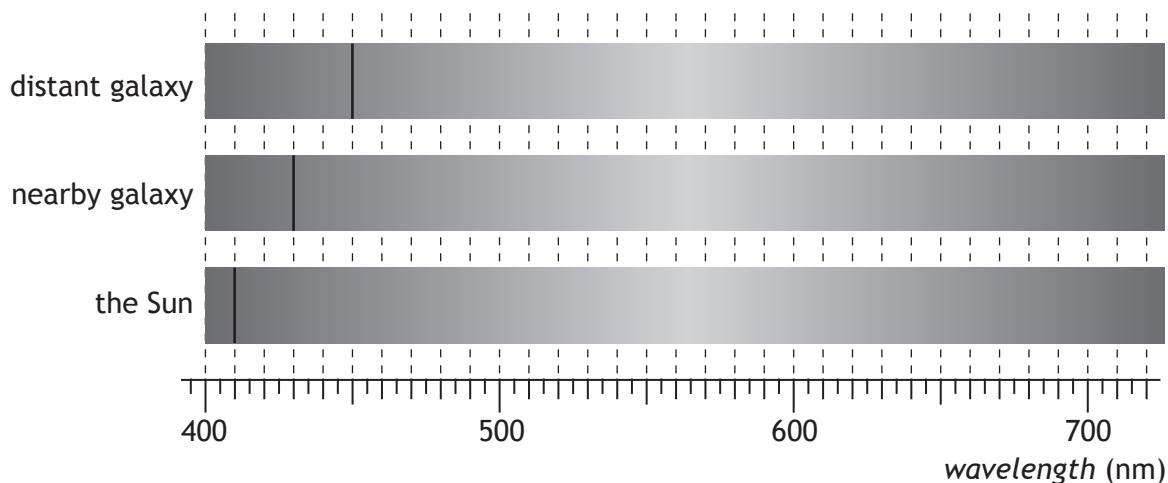
(i) Explain why the position of the line is different in each of the spectra.　　**2**

4. (b) (continued)

(ii) Show that the redshift of the light from the distant galaxy is 0·098. **2**

Space for working and answer

(iii) Calculate the approximate distance to the distant galaxy. **5**

Space for working and answer

[Turn over

5. A quote from a well-known science fiction writer states:

"In the beginning there was nothing, which exploded."

Using your knowledge of physics, comment on the above statement. **3**

6. (a) The Standard Model classifies *force mediating particles* as bosons. Name the boson associated with the electromagnetic force. **1**

(b) In July 2012 scientists at CERN announced that they had found a particle that behaved in the way that they expected the Higgs boson to behave. Within a year this particle was confirmed to be a Higgs boson.

This Higgs boson had a mass-energy equivalence of 126 GeV.
(1 eV = $1 \cdot 6 \times 10^{-19}$ J)

  (i) Show that the mass of the Higgs boson is $2 \cdot 2 \times 10^{-25}$ kg. **3**

Space for working and answer

  (ii) Compare the mass of the Higgs boson with the mass of a proton in terms of orders of magnitude. **2**

Space for working and answer

[Turn over

7. The use of analogies from everyday life can help better understanding of physics concepts. Throwing different balls at a coconut shy to dislodge a coconut is an analogy which can help understanding of the photoelectric effect.

Use your knowledge of physics to comment on this analogy. **3**

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8. A student investigates how irradiance I varies with distance d from a point source of light.
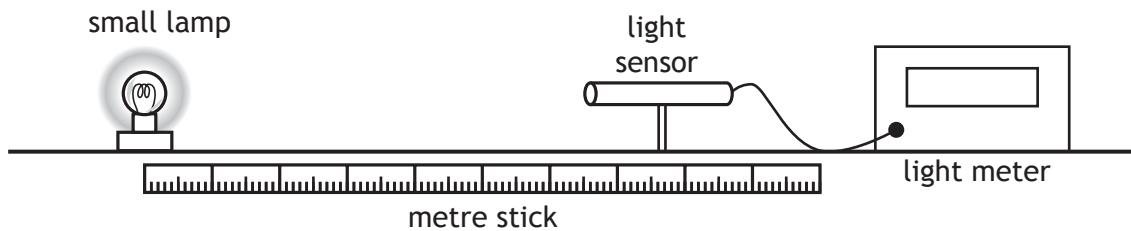
The distance between a small lamp and a light sensor is measured with a metre stick. The irradiance is measured with a light meter.

The apparatus is set up as shown in a darkened laboratory.

The following results are obtained.

d (m)	0·20	0·30	0·40	0·50
I (W m^{-2})	134·0	60·5	33·6	21·8

(a) State what is meant by the term *irradiance*. **1**

(b) Use **all** the data to establish the relationship between irradiance I and distance d . **3**

8. (continued)

(c) The lamp is now moved to a distance of 0·60 m from the light sensor.

Calculate the irradiance of light from the lamp at this distance. **3**

Space for working and answer

(d) Suggest one way in which the experiment could be improved.

You **must** justify your answer. **2**

(e) The student now replaces the lamp with a different small lamp.
The power output of this lamp is 24 W.

Calculate the irradiance of light from this lamp at a distance of 2·0 m. **4**

Space for working and answer

9. A student carries out two experiments to investigate the spectra produced from a ray of white light.

(a) In the first experiment, a ray of white light is incident on a glass prism as shown.

not to scale
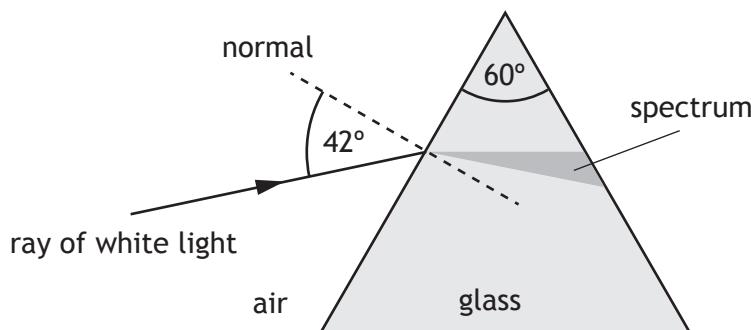
(i) Explain why a spectrum is produced in the glass prism. **1**

(ii) The refractive index of the glass for red light is 1·54.

Calculate the speed of red light in the glass prism. **3**

Space for working and answer

9.   (continued)

(b) In the second experiment, a ray of white light is incident on a grating.

not to scale
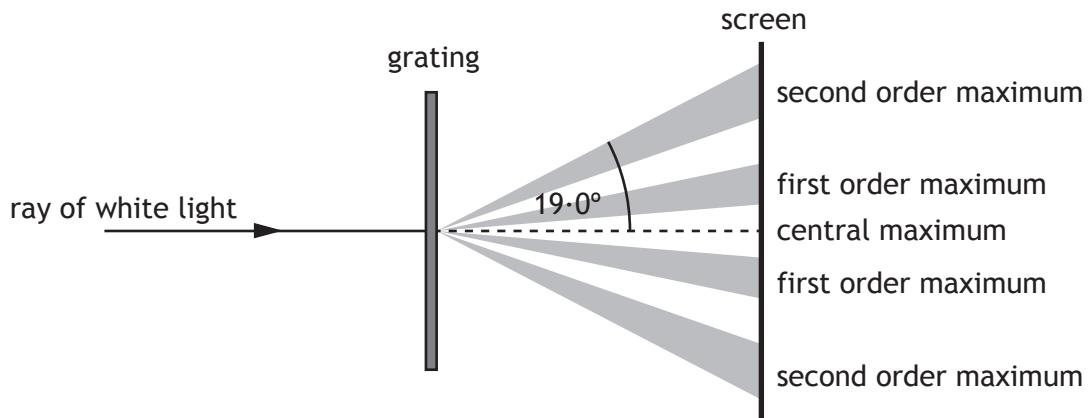
The angle between the central maximum and the second order maximum for red light is 19·0°.
The frequency of this red light is $4·57 \times 10^{14}$ Hz.

(i) Calculate the distance between the slits on this grating.   **5**

Space for working and answer

(ii) Explain why the angle to the second order maximum for blue light is different to that for red light.   **3**

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10. A car battery is connected to an electric motor as shown.
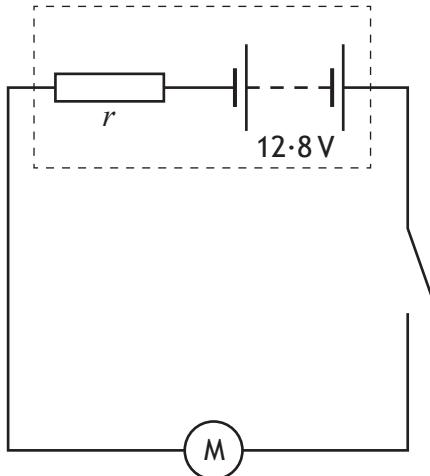
The electric motor requires a large current to operate.

(a) The car battery has an e.m.f. of 12·8 V and an internal resistance r of $6·0 \times 10^{-3}\,\Omega$. The motor has a resistance of $0·050\,\Omega$.

   (i) State what is meant by an *e.m.f. of 12·8 V*. **1**

   (ii) Calculate the current in the circuit when the motor is operating. **3**

     Space for working and answer

   (iii) Suggest why the connecting wires used in this circuit have a large diameter. **1**

10. (continued)

(b) A technician sets up the following circuit with a different car battery connected to a variable resistor R.
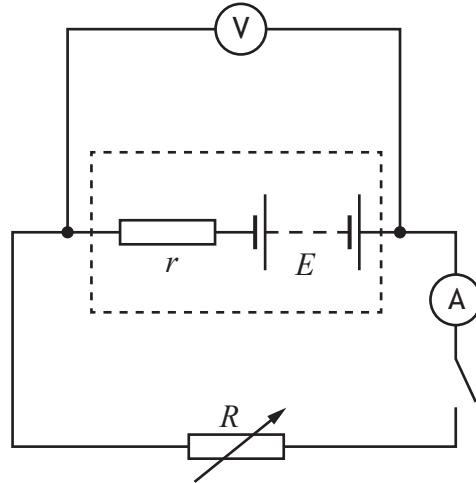
Readings of current I and terminal potential difference V from this circuit are used to produce the following graph.
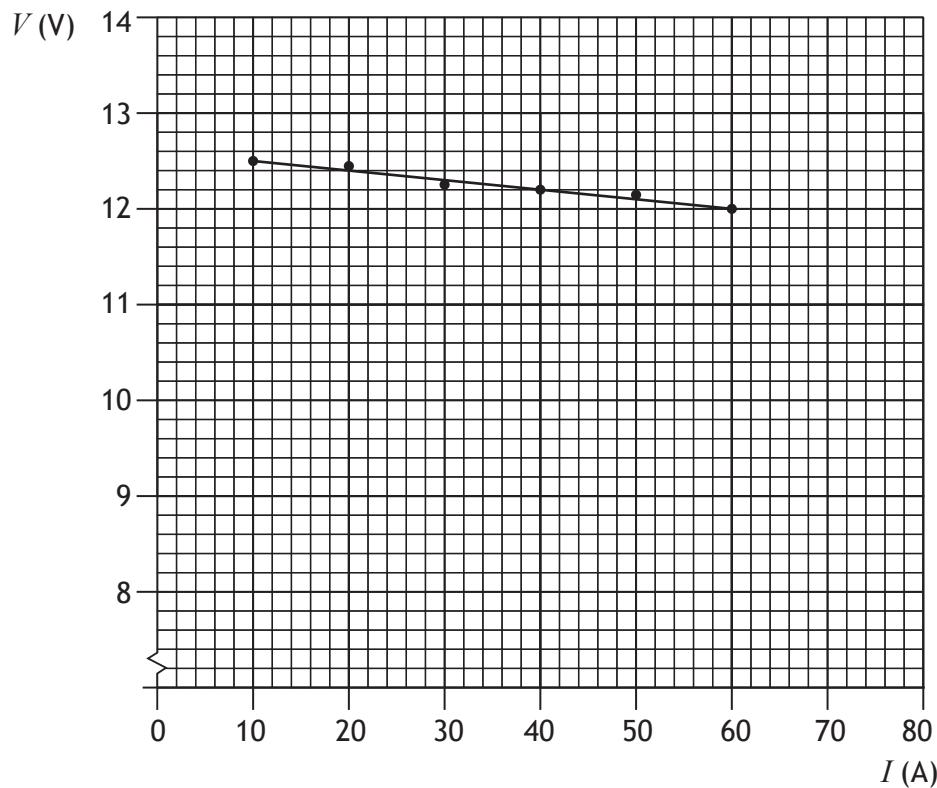
10. **(b)** **(continued)**

Use information from the graph to determine:

(i) the e.m.f. of the battery; **1**

Space for working and answer

(ii) the internal resistance of the battery; **3**

Space for working and answer

[Turn over

10. **(b)** **(continued)**

(iii) After being used for some time the e.m.f. of the battery decreases to 11·5 V and the internal resistance increases to 0·090 Ω.

The battery is connected to a battery charger of constant e.m.f. 15·0 V and internal resistance of 0·45 Ω as shown.
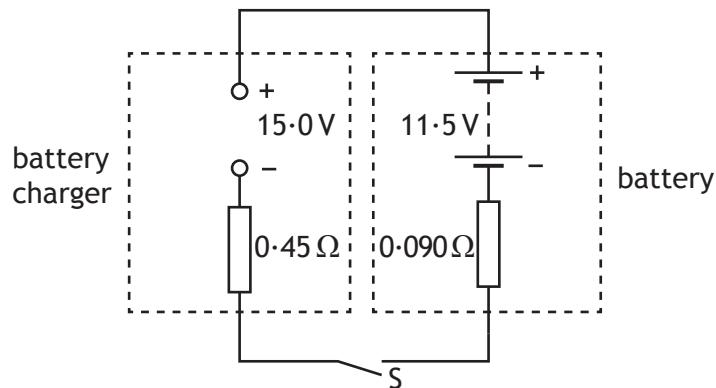
(A) Switch S is closed.

Calculate the initial charging current. **3**

Space for working and answer

(B) Explain why the charging current decreases as the battery charges. **2**

[Turn over for Question 11 on *Page thirty*

DO NOT WRITE ON THIS PAGE

11. A defibrillator is a device that provides a high energy electrical impulse to correct abnormal heart beats.

paddle A   paddle B
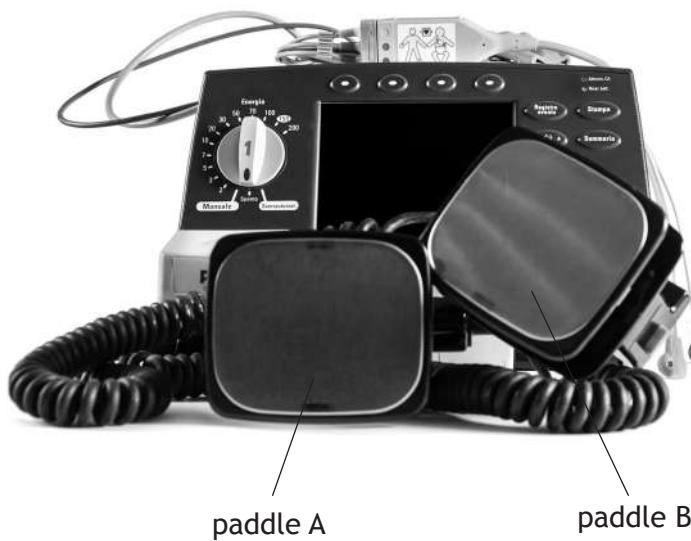
The diagram shows a simplified version of a defibrillator circuit.
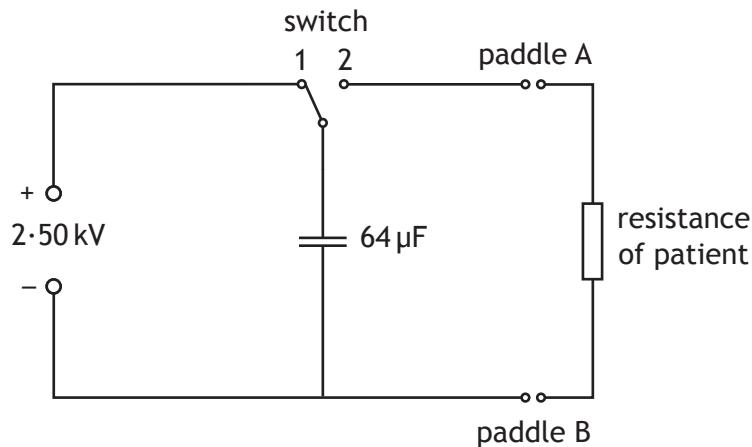
The switch is set to position 1 and the capacitor charges.

(a) Show the charge on the capacitor when it is fully charged is 0·16 C. **2**

Space for working and answer

11. **(continued)**

(b) Calculate the maximum energy stored by the capacitor. **3**

Space for working and answer

(c) To provide the electrical impulse required the capacitor is discharged through the person's chest using the paddles as shown

The initial discharge current through the person is 35·0 A.

(i) Calculate the effective resistance of the part of the person's body between the paddles. **3**

Space for working and answer

11. (c) (continued)

(ii) The graph shows how the current between the paddles varies with time during the discharge of the capacitor.
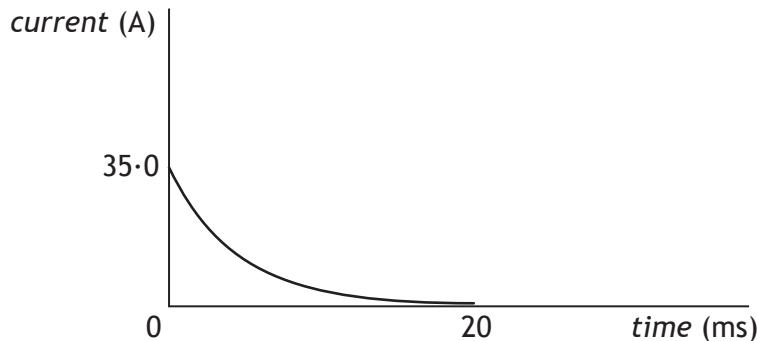
The effective resistance of the person remains the same during this time.

Explain why the current decreases with time. **1**

(iii) The defibrillator is used on a different person with larger effective resistance. The capacitor is again charged to 2·50 kV.

On the graph in (c)(ii) add a line to show how the current in this person varies with time.

(An additional graph, if required, can be found on *Page thirty-eight*). **2**

12. A student carries out an investigation to determine the refractive index of a prism.

A ray of monochromatic light passes through the prism as shown.

not to scale
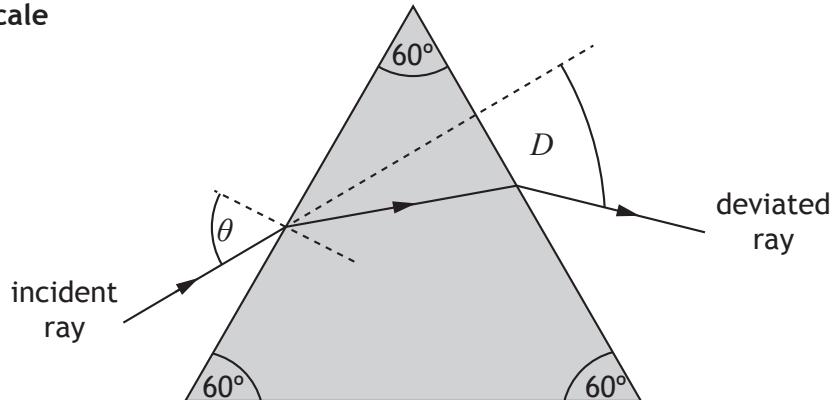
The angle of deviation D is the angle between the direction of the incident ray and the deviated ray.

The student varies the angle of incidence θ and measures the corresponding angles of deviation D.

The results are shown in the table.

Angle of incidence θ (°)	Angle of deviation D (°)
30·0	47·0
40·0	38·1
50·0	37·5
60·0	38·8
70·0	42·5

(a) Using the square-ruled paper on *Page thirty-five*, draw a graph of D against θ. **3**

(b) Using your graph state the two values of θ that produce an angle of deviation of 41·0°. **1**

(c) Using your graph give an estimate of the minimum angle of deviation D_m. **1**

12. **(continued)**

(d) The refractive index n of the prism can be determined using the relationship.

$$n \sin\left(\frac{A}{2}\right) = \sin\left(\frac{A + D_m}{2}\right)$$

where A is the angle at the top of the prism, and
D_m is the minimum angle of deviation.

Use this relationship and your answer to (c) to determine the refractive index of the prism.   **2**

Space for working and answer

(e) Using the same apparatus, the student now wishes to determine more precisely the minimum angle of deviation.

Suggest two improvements to the experimental procedure that would achieve this.   **2**

[END OF QUESTION PAPER]
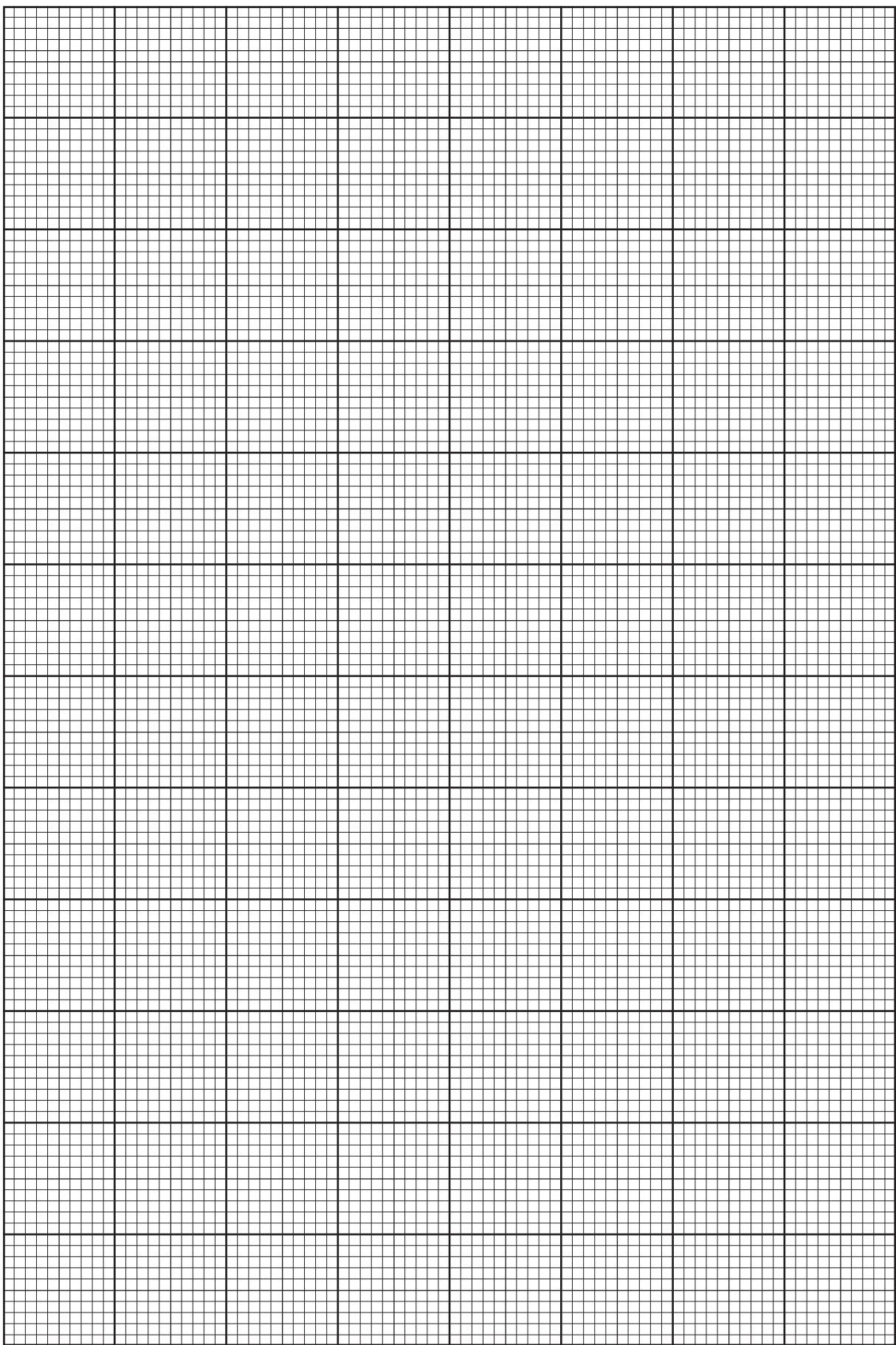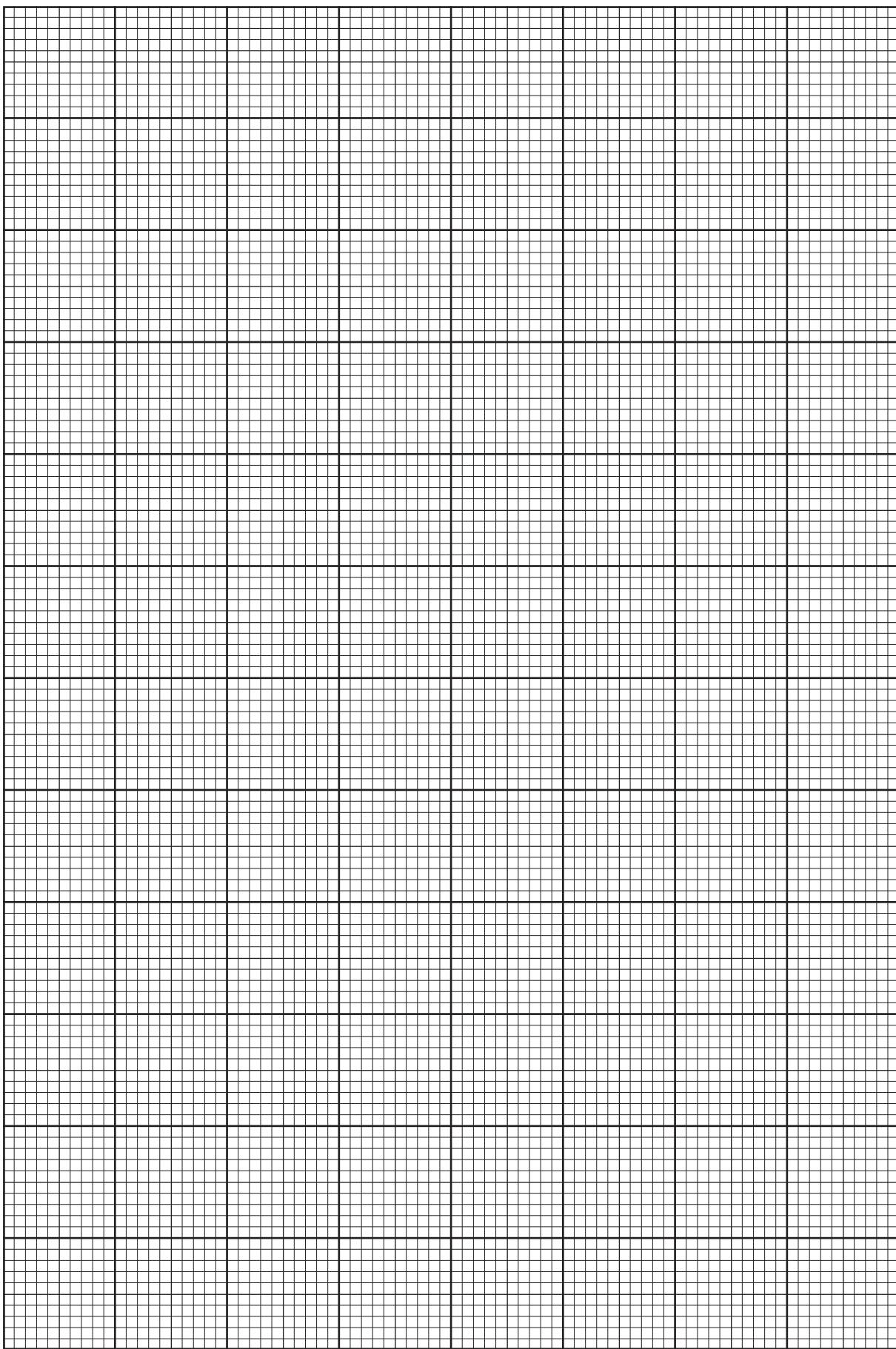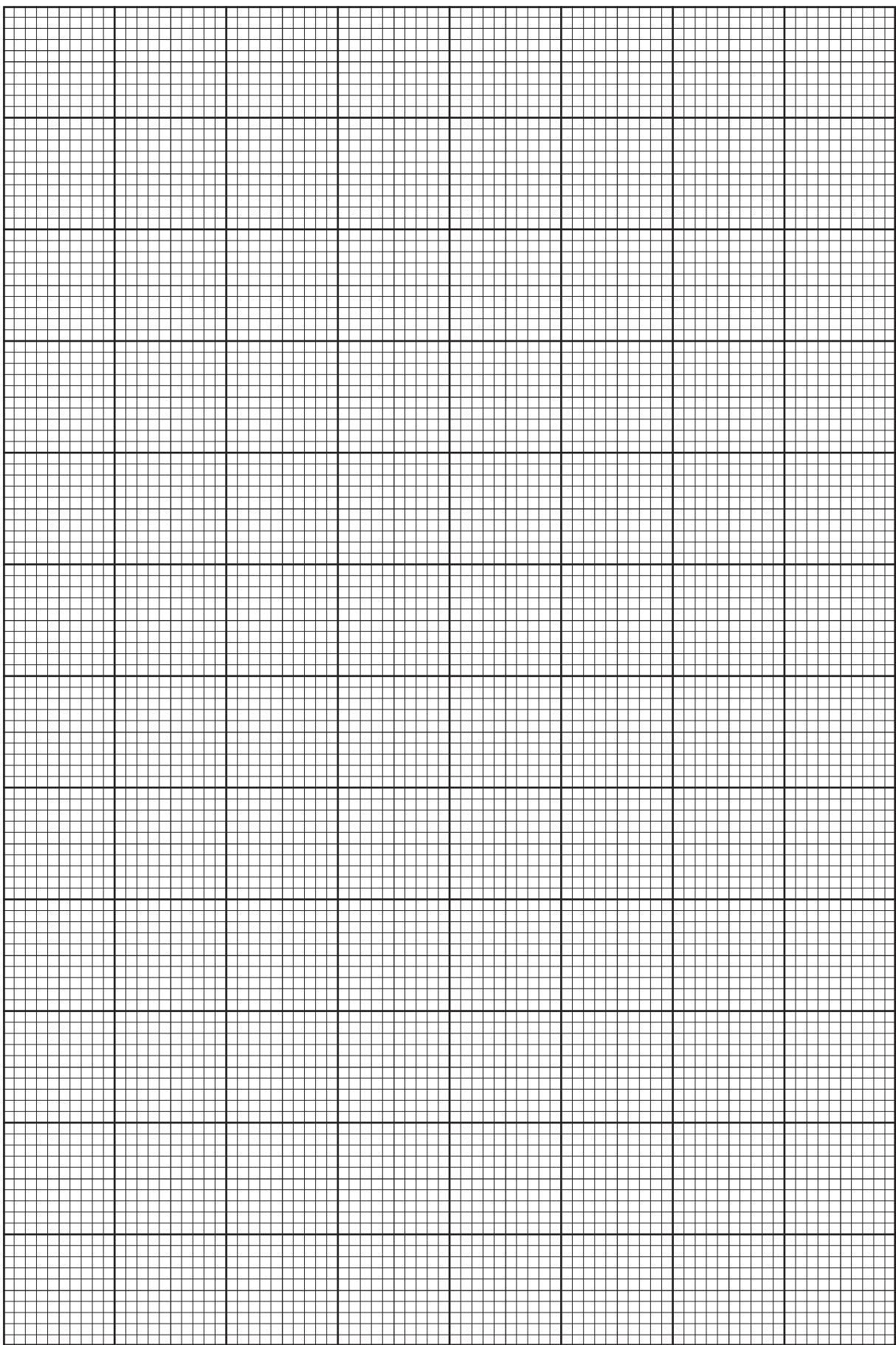
ADDITIONAL SPACE FOR ANSWERS AND ROUGH WORK

Additional graph for Question 11 (c)(iii)
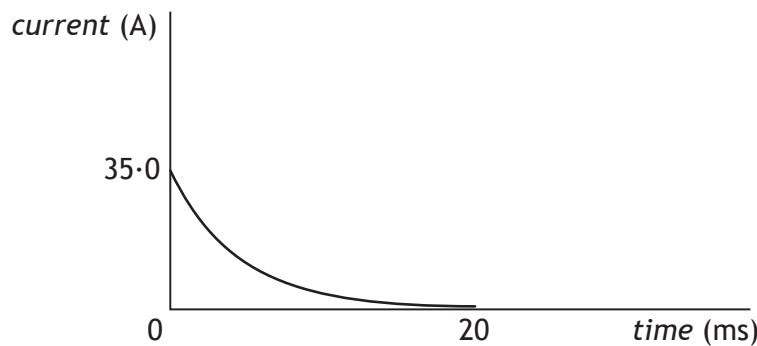
current (A)

35·0

0   20   *time* (ms)

Relationships required for Physics Higher

$d = \bar{v}t$

$s = \bar{v}t$

$v = u + at$

$s = ut + \frac{1}{2}at^2$

$v^2 = u^2 + 2as$

$s = \frac{1}{2}(u + v)t$

$W = mg$

$F = ma$

$E_W = Fd$

$E_p = mgh$

$E_k = \frac{1}{2}mv^2$

$P = \dfrac{E}{t}$

$p = mv$

$Ft = mv - mu$

$F = G\dfrac{m_1 m_2}{r^2}$

$t' = \dfrac{t}{\sqrt{1 - \left(\frac{v}{c}\right)^2}}$

$l' = l\sqrt{1 - \left(\frac{v}{c}\right)^2}$

$f_o = f_s\left(\dfrac{v}{v \pm v_s}\right)$

$z = \dfrac{\lambda_{observed} - \lambda_{rest}}{\lambda_{rest}}$

$z = \dfrac{v}{c}$

$v = H_0 d$

$W = QV$

$E = mc^2$

$E = hf$

$E_k = hf - hf_0$

$E_2 - E_1 = hf$

$T = \dfrac{1}{f}$

$v = f\lambda$

$d\sin\theta = m\lambda$

$n = \dfrac{\sin\theta_1}{\sin\theta_2}$

$\dfrac{\sin\theta_1}{\sin\theta_2} = \dfrac{\lambda_1}{\lambda_2} = \dfrac{v_1}{v_2}$

$\sin\theta_c = \dfrac{1}{n}$

$I = \dfrac{k}{d^2}$

$I = \dfrac{P}{A}$

path difference $= m\lambda$ or $\left(m + \frac{1}{2}\right)\lambda$ where $m = 0, 1, 2 \ldots$

random uncertainty $= \dfrac{\text{max. value} - \text{min. value}}{\text{number of values}}$

$V_{peak} = \sqrt{2}V_{rms}$

$I_{peak} = \sqrt{2}I_{rms}$

$Q = It$

$V = IR$

$P = IV = I^2R = \dfrac{V^2}{R}$

$R_T = R_1 + R_2 + \ldots$

$\dfrac{1}{R_T} = \dfrac{1}{R_1} + \dfrac{1}{R_2} + \ldots$

$E = V + Ir$

$V_1 = \left(\dfrac{R_1}{R_1 + R_2}\right)V_s$

$\dfrac{V_1}{V_2} = \dfrac{R_1}{R_2}$

$C = \dfrac{Q}{V}$

$E = \frac{1}{2}QV = \frac{1}{2}CV^2 = \frac{1}{2}\dfrac{Q^2}{C}$

Additional Relationships

Circle

circumference $= 2\pi r$

area $= \pi r^2$

Sphere

area $= 4\pi r^2$

volume $= \dfrac{4}{3}\pi r^3$

Trigonometry

$\sin \Theta = \dfrac{\text{opposite}}{\text{hypotenuse}}$

$\cos \Theta = \dfrac{\text{adjacent}}{\text{hypotenuse}}$

$\tan \Theta = \dfrac{\text{opposite}}{\text{adjacent}}$

$\sin^2 \Theta + \cos^2 \Theta = 1$

Electron Arrangements of Elements

Key

Field
Atomic number
Symbol
Electron arrangement
Name

Transition Elements

Group 1 (1)

Atomic number	Symbol	Electron arrangement	Name
1	H	(1)	Hydrogen
3	Li	2,1	Lithium
11	Na	2,8,1	Sodium
19	K	2,8,8,1	Potassium
37	Rb	2,8,18,8,1	Rubidium
55	Cs	2,8,18,18,8,1	Caesium
87	Fr	2,8,18,32,18,8,1	Francium

Group 2 (2)

Atomic number	Symbol	Electron arrangement	Name
4	Be	2,2	Beryllium
12	Mg	2,8,2	Magnesium
20	Ca	2,8,8,2	Calcium
38	Sr	2,8,18,8,2	Strontium
56	Ba	2,8,18,18,8,2	Barium
88	Ra	2,8,18,32,18,8,2	Radium

Group 3 (3)

Atomic number	Symbol	Electron arrangement	Name
21	Sc	2,8,9,2	Scandium
39	Y	2,8,18,9,2	Yttrium
57	La	2,8,18,18,9,2	Lanthanum
89	Ac	2,8,18,32,18,9,2	Actinium

(4)

Atomic number	Symbol	Electron arrangement	Name
22	Ti	2,8,10,2	Titanium
40	Zr	2,8,18,10,2	Zirconium
72	Hf	2,8,18,32,10,2	Hafnium
104	Rf	2,8,18,32,32,10,2	Rutherfordium

(5)

Atomic number	Symbol	Electron arrangement	Name
23	V	2,8,11,2	Vanadium
41	Nb	2,8,18,12,1	Niobium
73	Ta	2,8,18,32,11,2	Tantalum
105	Db	2,8,18,32,32,11,2	Dubnium

(6)

Atomic number	Symbol	Electron arrangement	Name
24	Cr	2,8,13,1	Chromium
42	Mo	2,8,18,13,1	Molybdenum
74	W	2,8,18,32,12,2	Tungsten
106	Sg	2,8,18,32,32,12,2	Seaborgium

(7)

Atomic number	Symbol	Electron arrangement	Name
25	Mn	2,8,13,2	Manganese
43	Tc	2,8,18,13,2	Technetium
75	Re	2,8,18,32,13,2	Rhenium
107	Bh	2,8,18,32,32,13,2	Bohrium

(8)

Atomic number	Symbol	Electron arrangement	Name
26	Fe	2,8,14,2	Iron
44	Ru	2,8,18,15,1	Ruthenium
76	Os	2,8,18,32,14,2	Osmium
108	Hs	2,8,18,32,32,14,2	Hassium

(9)

Atomic number	Symbol	Electron arrangement	Name
27	Co	2,8,15,2	Cobalt
45	Rh	2,8,18,16,1	Rhodium
77	Ir	2,8,18,32,15,2	Iridium
109	Mt	2,8,18,32,32,15,2	Meitnerium

(10)

Atomic number	Symbol	Electron arrangement	Name
28	Ni	2,8,16,2	Nickel
46	Pd	2,8,18,18,0	Palladium
78	Pt	2,8,18,32,17,1	Platinum
110	Ds	2,8,18,32,32,17,1	Darmstadtium

(11)

Atomic number	Symbol	Electron arrangement	Name
29	Cu	2,8,18,1	Copper
47	Ag	2,8,18,18,1	Silver
79	Au	2,8,18,32,18,1	Gold
111	Rg	2,8,18,32,32,18,1	Roentgenium

(12)

Atomic number	Symbol	Electron arrangement	Name
30	Zn	2,8,18,2	Zinc
48	Cd	2,8,18,18,2	Cadmium
80	Hg	2,8,18,32,18,2	Mercury
112	Cn	2,8,18,32,32,18,2	Copernicium

Group 3 (13)

Atomic number	Symbol	Electron arrangement	Name
5	B	2,3	Boron
13	Al	2,8,3	Aluminium
31	Ga	2,8,18,3	Gallium
49	In	2,8,18,18,3	Indium
81	Tl	2,8,18,32,18,3	Thallium

Group 4 (14)

Atomic number	Symbol	Electron arrangement	Name
6	C	2,4	Carbon
14	Si	2,8,4	Silicon
32	Ge	2,8,18,4	Germanium
50	Sn	2,8,18,18,4	Tin
82	Pb	2,8,18,32,18,4	Lead

Group 5 (15)

Atomic number	Symbol	Electron arrangement	Name
7	N	2,5	Nitrogen
15	P	2,8,5	Phosphorus
33	As	2,8,18,5	Arsenic
51	Sb	2,8,18,18,5	Antimony
83	Bi	2,8,18,32,18,5	Bismuth

Group 6 (16)

Atomic number	Symbol	Electron arrangement	Name
8	O	2,6	Oxygen
16	S	2,8,6	Sulfur
34	Se	2,8,18,6	Selenium
52	Te	2,8,18,18,6	Tellurium
84	Po	2,8,18,32,18,6	Polonium

Group 7 (17)

Atomic number	Symbol	Electron arrangement	Name
9	F	2,7	Fluorine
17	Cl	2,8,7	Chlorine
35	Br	2,8,18,7	Bromine
53	I	2,8,18,18,7	Iodine
85	At	2,8,18,32,18,7	Astatine

Group 0 (18)

Atomic number	Symbol	Electron arrangement	Name
2	He	2	Helium
10	Ne	2,8	Neon
18	Ar	2,8,8	Argon
36	Kr	2,8,18,8	Krypton
54	Xe	2,8,18,18,8	Xenon
86	Rn	2,8,18,32,18,8	Radon

Lanthanides

Atomic number	Symbol	Electron arrangement	Name
57	La	2,8,18,18,9,2	Lanthanum
58	Ce	2,8,18,20,8,2	Cerium
59	Pr	2,8,18,21,8,2	Praseodymium
60	Nd	2,8,18,22,8,2	Neodymium
61	Pm	2,8,18,23,8,2	Promethium
62	Sm	2,8,18,24,8,2	Samarium
63	Eu	2,8,18,25,8,2	Europium
64	Gd	2,8,18,25,9,2	Gadolinium
65	Tb	2,8,18,27,8,2	Terbium
66	Dy	2,8,18,28,8,2	Dysprosium
67	Ho	2,8,18,29,8,2	Holmium
68	Er	2,8,18,30,8,2	Erbium
69	Tm	2,8,18,31,8,2	Thulium
70	Yb	2,8,18,32,8,2	Ytterbium
71	Lu	2,8,18,32,9,2	Lutetium

Actinides

Atomic number	Symbol	Electron arrangement	Name
89	Ac	2,8,18,32,18,9,2	Actinium
90	Th	2,8,18,32,18,10,2	Thorium
91	Pa	2,8,18,32,20,9,2	Protactinium
92	U	2,8,18,32,21,9,2	Uranium
93	Np	2,8,18,32,22,9,2	Neptunium
94	Pu	2,8,18,32,24,8,2	Plutonium
95	Am	2,8,18,32,25,8,2	Americium
96	Cm	2,8,18,32,25,9,2	Curium
97	Bk	2,8,18,32,27,8,2	Berkelium
98	Cf	2,8,18,32,28,8,2	Californium
99	Es	2,8,18,32,29,8,2	Einsteinium
100	Fm	2,8,18,32,30,8,2	Fermium
101	Md	2,8,18,32,31,8,2	Mendelevium
102	No	2,8,18,32,32,8,2	Nobelium
103	Lr	2,8,18,32,32,9,2	Lawrencium